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ASYMMETRIC SYNTHESIS OF UNUSUAL AMINO ACIDS: SYNTHESIS OF OPTICALLY PURE ISOMERS OF &-METHYLTYROSINE

Ernesto Nicolas, Ramalinga Dharanipragada, Geza Toth and Victor J. Hruby*, Department of Chemistry, University of Arizona, Tucson, AZ 85721 U.S.A.

Abstract: Synthesis of optically pure isomers of ß-methyltyrosine has been accomplished.

In the preceeding letter, we have detailed our approach to the synthesis of optically pure isomers of ß-methylphenylalanine in very high optical purities, using in part the chiral imide enolate bromination methodology of Evans & coworkers.¹ Following protocols that are similar, we have synthesized the optically pure isomers of ß-methyltyrosine. In addition, we have converted erythro-D,L-ß-methylphenylalanine to erythro-D,L-ß-methyltyrosine. The details of these syntheses are described in this letter.²

To obtain optically pure isomers of ß-methyltyrosine following protocols described in the preceeding paper, we needed chiral auxiliary $\underline{1}$ and (+) or (-)4'-methoxy-3-phenylbutyric acid 2 (Scheme-1),



Since the desired acid $\underline{2}$ is not commercially available, we synthesized it from 4-methoxyacetophenone using the Wittig reaction (Scheme-2). Utilizing potassium t-butoxide (instead of sodium hydride) as base and tetrahydrofuran as solvent (instead of diglyme) improved the reported³ yields of $\underline{4}$ to 97%. The α,β -unsaturated ester $\underline{4}$, was subjected to catalytic hydrogenation (10% pd/c), followed by hydrolysis to give the racemic acid $\underline{2}$ in 92% overall yield. It was resolved via fractional crystallization of diastereoisomeric salts formed with S-(-)- and R-(+)- α -methylbenzylamine, respectively. The absolute configuration of (+) and (-)4'-methoxy-3-phenylbutyric acid is not known and we have determined (Scheme-3) that the (+) acid has the S configuration by chemical correlation with the known⁴ methyl ester of S-(+)-3-phenylburyricacid $\underline{2}$. Thus, (+)-4'-methoxy-3-phenylbutyric acid $\underline{2}$ was demethylated⁵ and esterified, but attempted deoxygenation⁶ of the corresponding triflate $\underline{8}$ with triethylammoniumformate in the presence of a homogenous palladium catalyst⁷ failed. The phenol $\underline{5}$ was converted⁸ into tetrazole $\underline{6}$, followed by reductive removal of the tetrazolloxy functionality using catalytic transfer hydrogenation gave (+)methyl 3-phenylbutyrate ($[\alpha]^{23}_{545}$ + 74.4° (c 1.0, benzene)), identical in all respects (IR, NMR, MS) to an authentic sample ($[\alpha]^{23}_{546}$ + 75° (c 1.0, benzene)) of the ester prepared from S-(+)-3-phenylbutyric acid.







For the synthesis of β -methyltyrosine, S-(+)-4'-methoxy-3-phenylbutyric acid was coupled with (Scheme-4) the chiral auxiliary derived from L-phenylalanine⁹ to give the Nacyloxazolidinone <u>10</u>. Enolate bromination followed by displacement with azide gave <u>12</u>. Removal of the chiral auxiliary, followed by reduction of the azido group and hydrolysis of methyl ether gave erythro(2R,3R)-D- β -methyltyrosine in 98% d.e. as judged by ¹H NMR. Utilizing R-(-)-4'-methoxy-3-phenylbutyric acid and the L-chiral auxiliary gave threo(2R,3S)-D- β methyltyrosine in 94% d.e. The synthesis of the other two isomers using the chiral auxiliary derived from D-phenylalanine is now in progress. We have synthesized racemic erythro- β methyltyrosine from racemic erythro- β -methylphenylalanine via nitration, reduction, diazotization and hydroxylation (The racemic material was identical in all respects except for optical rotation to the material obtained by asymmetric synthesis.). In conclusion, unusual amino acids such as β -methyltyrosine can be readily synthesized¹⁰ in high optical purities, utilizing in part, the chiral imide enolate bromination methodology.¹ Acknowledgements: This work was supported by funds from the National Science Foundation and the United States Public Health Service. We thank Mr. Casey Russell, Mr. Soaring Bear, Ms. Anne Bannister and Ms. Lisa Kennedy for technical assistance. E.N. thanks C.I.R.I.T. (Spain) for financial support.

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